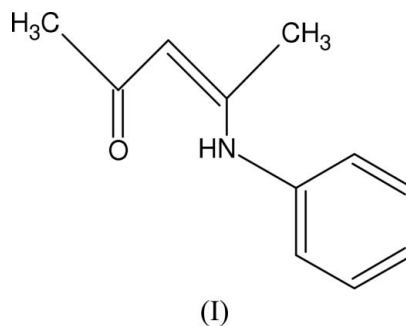


**(Z)-4-Anilinopent-3-en-2-one**Farkhanda Shaheen,<sup>a\*</sup> L Marchio,<sup>b</sup> Amin Badshah<sup>a</sup> and Muhammad Kaleem Khosa<sup>c</sup><sup>a</sup>Department of Chemistry, Quaid-i-Azam University, Islamabad 45320, Pakistan, <sup>b</sup>Dip. di Chimica Generale ed Inorganica Chimica Analitica Fisica, Università di Parma, Parco Area delle Scienze 17/A43100, Parma, Italy, and <sup>c</sup>Department of Chemistry, Quaid-i-Azam University, Islamabad 45320, Pakistan

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**Key indicators**Single-crystal X-ray study  
*T* = 293 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$   
*R* factor = 0.048  
*wR* factor = 0.139  
Data-to-parameter ratio = 11.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title compound,  $\text{C}_{11}\text{H}_{13}\text{NO}$ , crystallizes as the *Z* isomer of the  $\beta$ -enamino-ketone. An intramolecular hydrogen-bonding interaction exists between the N–H and C=O groups.Received 3 January 2006  
Accepted 26 January 2006**Comment**

The chelating tendencies of  $\beta$ -diketones have been extensively studied because of their applications in pharmaceutical compounds, biochemistry, biomedical research and immunochemistry (Sandler & Karo, 1986; Chen & Rhodes, 1996) and as precursors for the preparation of a large number of heterocyclic compounds (Khosropour *et al.*, 2004). Previous studies of some cyclizations involved the preparation of 4-methylaminopent-3-en-2-ones (Kascheres *et al.*, 2001), 5-(1-methylethylaminomethylene)-1,3-dioxane-4,6-diones and 5-(dimethylethylaminomethylene)-2,2-dimethyl-1,3-dioxane-4,6-diones (Zhuo, 1997). The anticonvulsant activities of various acyclic and cyclic enamino-ketones have been studied in connection with their lipophilicity, steric, electronic and hydrogen-bonding effects (Edafiogho *et al.*, 1994; Hinko *et al.*, 1993). The condensation products of acetylacetone with various amines exist in three forms, *viz.* the Schiff base, the ketamine and the enimine. The interchange between the ketamine and enimine tautomers involves a small displacement in the equilibrium position of the acidic proton.



In the title compound, (I), the C2=O1 and C2–C3 bond distances [1.224 (3) and 1.424 (4) Å, respectively] confirm the existence of the enamino-ketone. The bond distances in the C3=C4–N1 chain indicate greater electron delocalization [C3=C4 = 1.365 (3) Å and N1–C4 = 1.365 (3) Å]. The enamino-ketone fragment (N1–C4=C3–C2=O1) is essentially planar, the maximum deviation being 0.011 (3) Å for atom C2. The dihedral angle between this fragment and the phenyl ring is 32.06 (9)°. This is a result of the steric hindrance between the C5-methyl group and the phenyl ring, which hinders conjugation between the two groups. The

molecule is the *Z* isomer and an intramolecular hydrogen bond is present between the keto and imino groups [ $N1-H1 = 0.89$  (3) Å,  $H1 \cdots O1 = 1.92$  (3) Å,  $N1 \cdots O1 = 2.675$  (3) Å and  $N1-H1 \cdots O1 = 140$  (2)°].

### Experimental

The title compound was prepared by modification of a published method (Fustero *et al.*, 1999). Aniline (15.26 ml, 0.1 mol) was added to acetylacetone (20 ml, 0.1 mol) and two drops of  $H_2SO_4$  in benzene (50 ml). The reaction mixture was refluxed for 4 h. On cooling, the product was filtered and recrystallized from a chloroform/*n*-hexane (9:1 *v/v*) mixture.

#### Crystal data

$C_{11}H_{13}NO$	$D_x = 1.183 \text{ Mg m}^{-3}$
$M_r = 175.22$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 24 reflections
$a = 9.157$ (3) Å	$\theta = 6.0\text{--}17.8^\circ$
$b = 11.040$ (7) Å	$\mu = 0.08 \text{ mm}^{-1}$
$c = 10.130$ (5) Å	$T = 293$ (2) K
$\beta = 106.10$ (3)°	Prism, colorless
$V = 983.9$ (9) Å <sup>3</sup>	$0.60 \times 0.40 \times 0.40 \text{ mm}$
$Z = 4$	

#### Data collection

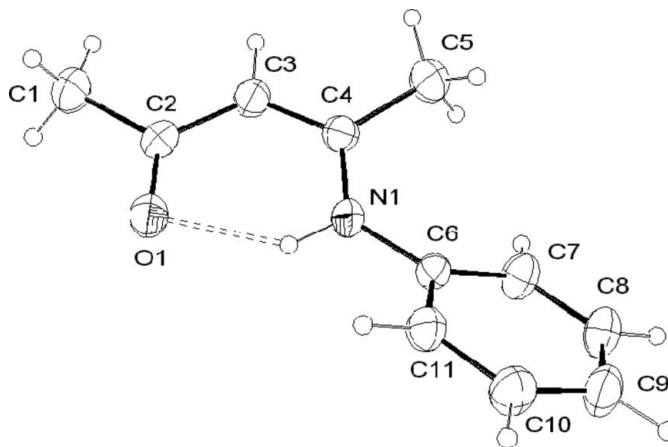
Philips PW 1100 diffractometer	$R_{\text{int}} = 0.022$
$\omega/2\theta$ scans	$\theta_{\text{max}} = 25.0^\circ$
Absorption correction: refined from $\Delta F$ [local program based on Walker & Stuart (1983)]	$h = -10 \rightarrow 10$
$T_{\text{min}} = 0.893$ , $T_{\text{max}} = 0.970$	$k = 0 \rightarrow 13$
1812 measured reflections	$l = 0 \rightarrow 12$
1717 independent reflections	1 standard reflections
1120 reflections with $I > 2\sigma(I)$	every 100 reflections
	intensity decay: none

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.059P)^2 + 0.3446P]$
$R[F^2 > 2\sigma(F^2)] = 0.048$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.139$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.16 \text{ e \AA}^{-3}$
1717 reflections	$\Delta\rho_{\text{min}} = -0.12 \text{ e \AA}^{-3}$
148 parameters	
H atoms treated by a mixture of independent and constrained refinement	

Methyl H atoms were located in difference Fourier syntheses and refined as rigid rotating groups, riding on their parent atoms, with  $C-H = 0.96$  Å and  $U_{\text{iso}}(\text{H}) = U_{\text{eq}}(\text{C})$ . All other H atoms were found in a difference map and refined freely.

Data collection: local program; cell refinement: local program; data reduction: local program; program(s) used to solve structure:



**Figure 1**

View of the structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. The dashed lines indicate the intramolecular hydrogen bond.

*SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *PARST* (Nardelli, 1995).

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